Compound Specific Challenges Associated with Trace Detection

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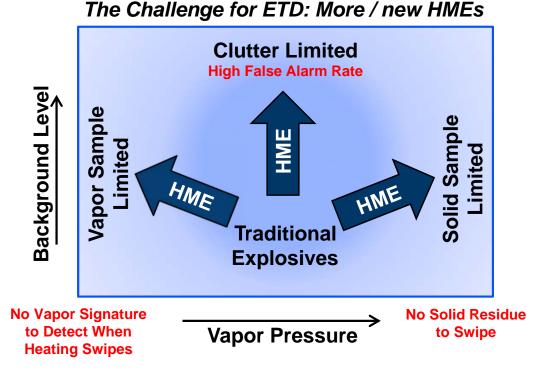


This work is sponsored by Department of Homeland Security, Science and Technology Directorate under Air Force Contract #FA8721-05-C-0002. Opinions, interpretations, recommendations and conclusions are those of the authors and are not necessarily endorsed by the United States Government.



- The explosive threat is evolving
 - More materials to detect
- Trace detection systems are evolving as well...
 - Improved sampling methods
 - Evolution from IMS to dual-polarity IMS to MS
 - See DHS S&T BAA 13-03 titled "Advanced Trace Detection Instrumentation and Methodologies"





MITLL is providing knowledge to help trace detection systems evolve

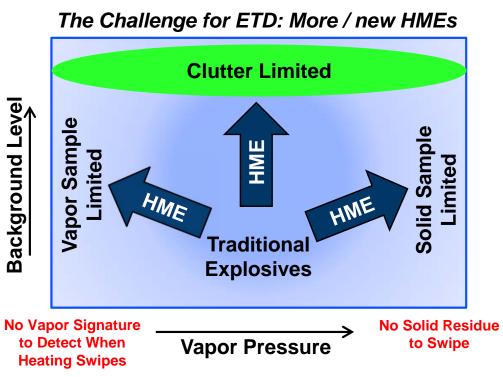


- Widely varying physical properties
 - Vapor pressures vary by up to eight orders of magnitude
 - Range of morphology (liquids, gels, crystalline solids, moldable plastics, machinable plastics, powders)
- Additional challenges
 - For some HMEs, their constituents may be present in the background environment



Focusing on detection of the main charge is not a single detection problem





- Challenge: HMEs constituent chemicals may be common in the background
- Case study: Ammonium nitrate
 - Common fertilizer
 - NH_3 (g) exists at ppb levels and HNO_3 (g) at ppt concentrations in the atmosphere
- AN exists in equilibrium with its precursors NH₃ and HNO₃

 NH_4NO_3 (s) $\Leftrightarrow NH_3$ (g) + HNO_3 (g)



ETD performance may be background limited

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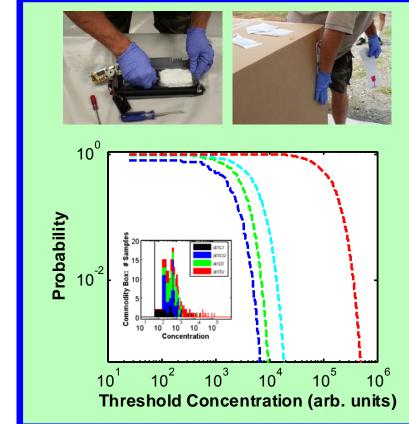
Method for Assessing Impact of Clutter

Background levels



- Background levels of salts similar for all cargo contents, cargo facility locations, seasons
- Simultaneous detection of nitrate and ammonium lowers
 P_{FA} by an order of magnitude

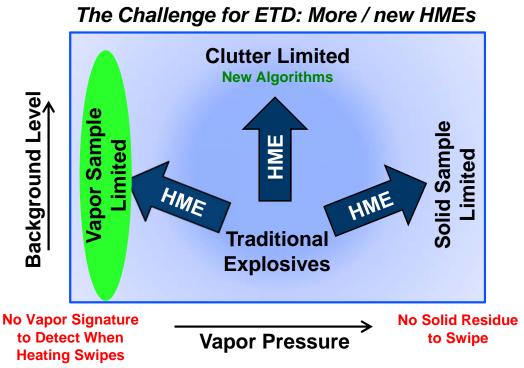
Threat Signatures



Examine clutter levels relative to signatures to set threshold requirements and identify algorithms/schemes for enhancing detection



ETD Challenge: Sample Desorption for Chlorates and Perchlorates



Perchiorate and Chiorate Menting and Boiling Point		
Compound	M.P. (°C)	B.P. (°C)
Potassium Perchlorate	525	600
Sodium Perchlorate	468	482
Potassium Chlorate	356	400
Sodium Chlorate	248	300
Perchloric Acid	-17	203
Chloric Acid	?	40

Perchlorate and Chlorate Melting and Boiling Points

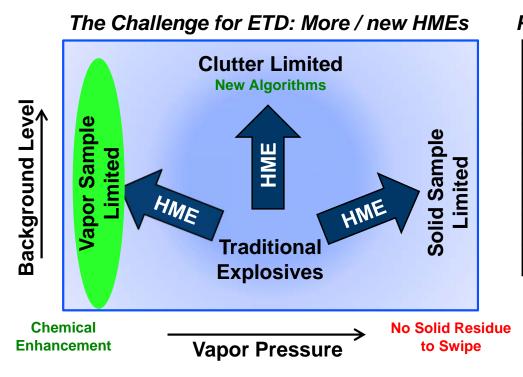
Industrial preparation: Salt Metathesis Reactions NaClO₄ + HCl \rightarrow NaCl + HClO₄ Ba(ClO₃)₂ + H₂SO₄ \rightarrow 2HClO₃ + BaSO₄

- High melting and boiling points translate into <u>low</u> vapor pressures at typical thermal desorption (TD) temperatures, 150 – 200 °C
- Low vapor pressures of chlorates and perchlorates limits TD based ETD
- However, perchloric acid (HCIO₄) and chloric acid (HCIO₃) have relatively high vapor pressures

Convert chlorate and perchlorate salts into chloric and perchloric acids to enable thermal desorption based detection strategies



ETD Challenge: Sample Desorption for Chlorates and Perchlorates



Perchlorate and Chlorate Melting and Boiling Points **B.P. (°C)** M.P. (°C) Compound Potassium Perchlorate 525 600 Sodium Perchlorate 482 468 Potassium Chlorate 356 400 Sodium Chlorate 300 248 **Perchloric Acid** -17 203 **Chloric Acid** 40 ?

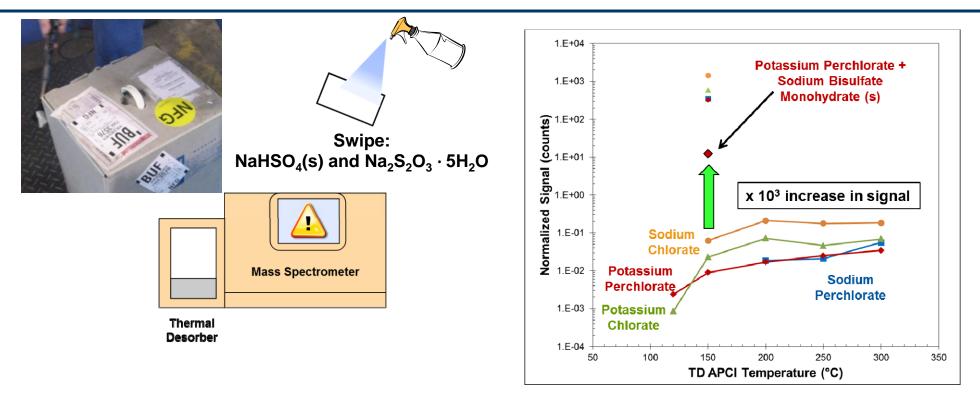
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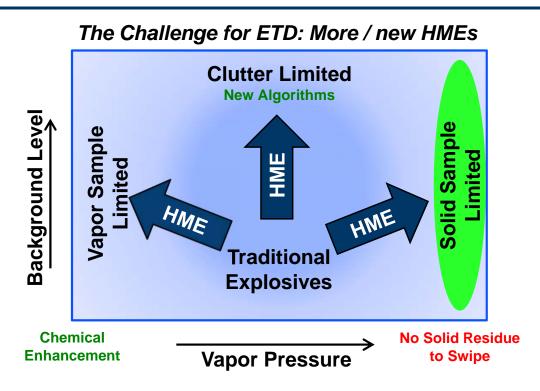
Acid Enhanced Detection of Chlorates and Perchlorates



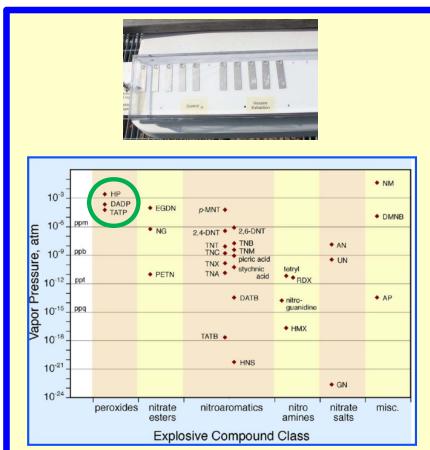
- Solution:
 - Add sodium bisulfate (NaHSO₄(s)); source of acidic protons and the sulfate anion
 - Codeposit sodium thiosulfate (Na₂S₂O₃ \cdot 5H₂O) as a 'dry' source of water upon thermal desorption
- **Pros:** Easy handling of dry swipes, desired chemistry is thermally activate
 - Sodium bisulfate and thiosulfate are safe

Dry swipe with embedded safe solid compounds will enhance detection of chlorates and perchlorates

ETD Challenge: High Vapor Pressure Regime



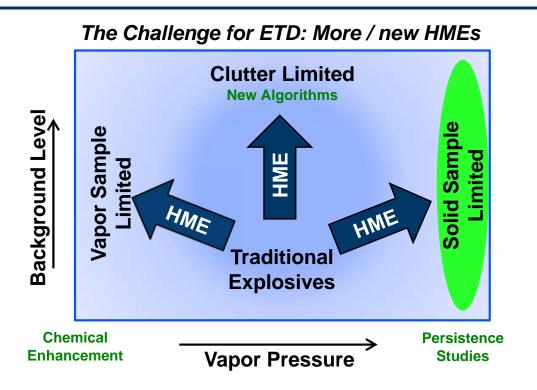
- TATP has a high vapor pressure
- Traditional sampling methods
 may need to be updated



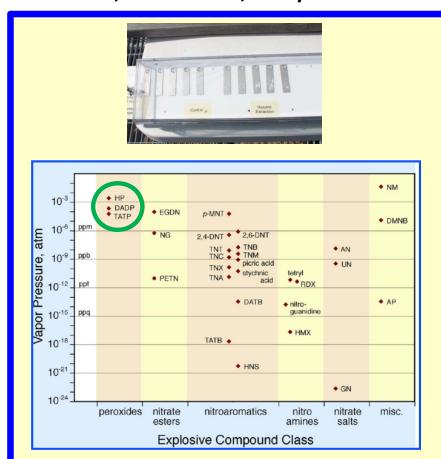
Fate, Persistence, Composition

Robert G. Ewing, Melanie J. Waltman, David A. Atkinson, Jay W. Grate, Peter J. Hotchkiss, TrAC Trends in Analytical Chemistry, Volume 42, January 2013, Pages 35-48.

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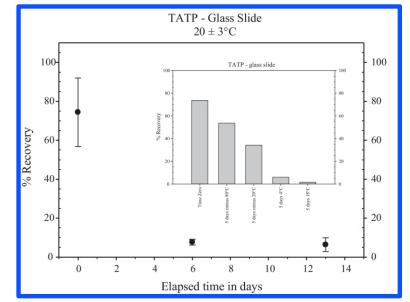


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ETD Challenge: Persistence of Residues





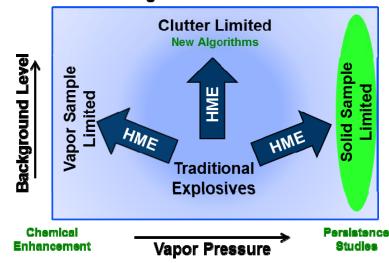
Nopporn Song-im, Sarah Benson, Chris Lennard, Forensic Science International, Volume 226, Issues 1–3, 10 March 2013, Pages 244-253

- TATP residues evaporate/decompose
 - Recent study showed that after 12 days at 20°C, 5% of TATP deposited on a glass slide remained (approximately ~1 μ g still available for detection)
 - Current MITLL work aimed at assessing requisite detection thresholds
- Additional Detection Mitigation Measures
 - Vapor sensor
 - Detect presence of associated chemicals (main charge, or decomposition products)
 - Bulk screening



- Chemical diversity of IEDs presents challenges for ETDs
- These challenges are being met two ways:
 - Improved instrumentation (industry focus)
 - Increased knowledge of trace phenomenology and background levels, leading to new methods and algorithms (MIT LL focus)
- The evolution of ETDs will ensure their future role in our counter-explosives architecture for air cargo security





The Challenge for ETD: More / new HMEs



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